Strictly Confidential- (For Internal and Restricted Use Only) Secondary School Examination SUMMATIVE ASSESSMENT - II July 2017

Marking Scheme – Science (for Blind Candidates) 31/B

- 1. The Marking Scheme provides general guidelines to reduce subjectivity in the marking. It carries only suggested value points for the answer. <u>These are only guidelines and do not constitute the complete answer</u>. Any other individual response with suitable justification should also be accepted even if there is no reference to the text.
- 2. Evaluation is to be done as per instructions provided in the Marking Scheme. It should not be done according to one's own interpretation or any other consideration. Marking Scheme should be strictly adhered to and religiously followed.
- 3. If a question has parts, please <u>award marks in the right hand side for each part</u>. Marks awarded for different parts of the question should then be totalled up and written in the left hand margin.
- 4. If a question does not have any parts, marks be awarded in the left hand side margin.
- 5. If a candidate has attempted an extra question, <u>marks obtained in the question attempted first</u> <u>should be retained</u> and the other answer should be scored out.
- 6. Wherever only two/three of a 'given' number of examples/factors/points are expected only the first two/three or expected number should be read. The rest are irrelevant and should not be examined.
- 7. There should be <u>no effort at 'moderation' of the marks</u> by the evaluating teachers. The actual total marks obtained by the candidate may be of no concern of the evaluators.
- 8. All the Head Examiners / Examiners are instructed that while evaluating the answer scripts, if the answer is found to be totally incorrect, the (X) should be marked on the incorrect answer and awarded '0' marks.
- 9. ½ mark may be deducted if a candidate either does not write units or writes wrong units in the final answer of a numerical problem.
- 10. A full scale of mark 0 to 100 has to be used. <u>Please do not hesitate to award full marks if the answer deserves it</u>.
- 11. As per orders of the Hon'ble Supreme Court the candidates would now be permitted to obtain photocopy of the Answer Book on request on payment of the prescribed fee. All Examiners/Head Examiners are once again reminded that they must <u>ensure that evaluation is carried out strictly</u> <u>as per value points given in the marking scheme</u>.

MARKING SCHEME CLASS X – BLIND

	Expected Answer/ Value point	Marks	Total
	SECTION – A		
Q 1.	Methanol, Ethanol	1/2 , 1/2	1
Q2.	Multiple fission, Asexual	1/2 , 1/2	1
Q3.	Where the living organism in an area interact with the non-living constituents of the environment.		1
Q4.	 Virtual Erect Magnified Behind the mirror Laterally inverted (Any four) 	½ x 4	2
Q5.	 Management of natural resources so that they meet the current basic human needs and last for the generations to come without being exploited for the short term gains. Necessary for the following reasons Ensure equitable distribution of resources To prevent environmental damage. Last for a longer duration and provide steady growth. 	1 1⁄2 x 2	2
Q6.	 A. K. Banerjee, a forest officer revived the badly degraded sal forests with the help/ involvement of villagers in West Bengal In return for help in protection, the villagers were given employment in both silviculture and harvesting operations Given 25% of final harvest Allowed to take fuelwood and fodder collection on the payment of a nominal fee. 	¹ ∕2 x 4	2
Q7.	• $C_{3}H_{6}$; because it is an unsaturated compound • $H H H$ R C = C R M Ni / Pd R C = C R R M R - C - C - R R R R • Used for hydrogenation of vegetable oils	1	3
	Used for hydrogenation of vegetable oils	1	3

Q8.	$A : C_2H_5OH / Ethanol$		
	B : CH ₃ COOH / Ethanoic Acid		
	$X: H_2$	¹∕₂ x 3	
	$CH_3 - CH_2OH \longrightarrow CH_3COOH$ acidified $K_2Cr_2O_7 + Heat$	1/2	
	$2CH_3COOH + 2Na \rightarrow 2CH_3COONa + H_2$	1	3
Q9.	 To make the study of elements systematic and easier Atomic number Modern periodic law "Properties of elements are a periodic function of their atomic number." 18 Groups, 7 Periods. 		3
Q10.	 a) Na / Sodium, 2,8,1 b) Group number- 1, Period number- 3 c) Sodium hydroxide, NaOH 	$\frac{1/2}{1/2}, \frac{1/2}{1/2}, \frac{1/2}{1/2}, \frac{1/2}{1/2}, \frac{1/2}{1/2}$	3
Q11.	• Transfer of pollen grains from anther to the stigma of a flower.	1	
~~~	• Self Pollination       Cross Pollination         1) Transfer of pollen grain to the stigma of the same flower or another flower of the same plant       1) Transfer of pollen grain to the stigma of another flower present on another plant of the same plant         2) Can occur only in bisexual flowers       2) Can occur in unisexual as well as bisexual flowers	1	3
Q12.	<ul> <li>Main steps of sexual reproduction;</li> <li>Formation of male and female gametes</li> <li>Meeting together of male and female gametes.</li> <li>Fusion of the two gametes to form a zygote.</li> <li>Development of zygote to form a new individual.</li> <li>Two advantages: <ol> <li>Brings about variation useful for the survival of species over time.</li> <li>Brings stability of population in a species.</li> </ol> </li> </ul>		3
Q13	<ul> <li>i) Pistil: 1) Receives the pollen grain</li> <li>2) Carries the male germ cell to the ovary</li> <li>3) Development of ovary and ovule into fruit and seeds         <ul> <li>(Any two)</li> <li>ii) Stamen: Produces pollen grains</li> </ul> </li> </ul>		3
Q14.	<ul> <li>When a cross was made between a tall pea plant with round seeds and a short pea plant with wrinkled seeds, the F1 progeny plants were all tall with round seeds; this indicates that tallness and round seeds are the dominant traits.</li> <li>When F1 plants were self pollinated, the F2 progeny consisted of some tall plants with round seeds and some short plants with wrinkled seeds</li> </ul>	1	

	which are the parental traits.	1		
	• There were also some new combinations like tall plants with wrinkled seeds and short plants with round seeds			
	• Thus it may be concluded that tall seeds traits have been inherited ind	/ short traits and round seed / wrinkled	1⁄2	
	OR A flow chart depicting the same			
	Note: Any other pair of contrasting characters can also be taken			3
Q15.	plan but are modified to perform different functions.		1	
	<ul> <li>Analogous organs: Organs having different origin and basic structural plan but are modified to perform a common function.</li> <li>Evidences: <ul> <li>The more characteristics any two species have in common, more closely they are related/ having a recent ancestor.</li> </ul> </li> </ul>			
	<ul> <li>Helps to identify an evolutio different species.</li> </ul>	nary relationship between apparently	1⁄2	3
Q16.	Real Image	Virtual Image		
Q10.	1) Can be obtained on a screen	1) Cannot be obtained on a screen		
	2) Image is inverted	2) Image is erect		
	3) Rays of light starting from a	3) Rays of light starting from a		
	common point actually converge	common point appear to diverge		
	after reflection/ refraction.	from a point after reflection/ refraction.		
		(any two)	1 x 2	
	<ul> <li>Characteristics of image -</li> <li>– Real</li> </ul>			
	– Inverted			
	– Magnified	( any two)	½ x 2	3
			-	-
Q17.	i) Myopia		1	
	Causes: 1) Elongation of the eye ball		1/2	
	2)Excessive curvature of the eye lens		1⁄2	
	ii) Nature: Concave Lens		1⁄2	
	Focal length: $-200 \text{ cm}$		1⁄2	
				3
Q18.	Biodegradable and non-biodegrada	ble	1/2 , 1/2	
×10.	<ul> <li>Saves time and energy</li> </ul>		$\frac{1}{2}, \frac{1}{2}$	
<ul> <li>Praveen is sincere, aware, responsib</li> </ul>		ble ( or any other)	$\frac{1}{2}, \frac{1}{2}$	3
	• I laveen is sincere, aware, responsi		, / 2	
Q19.	a) Isomers are compounds having same molecular formula but different structural formula.		1	
I 			I	

			-
	нннн	1	
	$\mathbf{H} - \dot{\mathbf{C}} - \dot{\mathbf{C}} - \dot{\mathbf{C}} - \mathbf{H}$		
	H H H H and		
	H = H = H = H = H = H = H = H = H = H =	1	
	b) Oxidation	1⁄2	
	$CH_3 - CH_2OH \xrightarrow{Alkaline KMnO_4 + Heat} CH_3COOH$	1⁄2	
	c) Ethene Cone, H.S.O., acts as the dehydrating agent	$\frac{1/2}{1/2}$	5
	Conc. $H_2SO_4$ acts as the dehydrating agent.	72	3
Q20.	a) Functions:		
<b>Q</b> ^{20.}	i) Ovary- Production of female gamete /egg		
	Production of female sex hormone/ oestrogen	$\frac{1}{2}, \frac{1}{2}$	
	ii) Uterus- Implantation of zygote/ fertilised egg	, _ , , _	
	Nourishment of the developing embryo	$\frac{1}{2}, \frac{1}{2}$	
	iii) Oviduct – Site of fertilization	,	
	Transfer of female gamete from ovary.	1/2 , 1/2	
	b) The embryo gets its nourishment inside the mother's body with the help of		
	a special tissue called placenta.	1/2	
	This is a disc embedded in the uterine wall which contains villi on the		
	embryo side of the tissue and blood spaces surrounding the villi on the	1	
	mother's side.	1/2	5
	It provides a large surface area for glucose and oxygen to pass from the mother to the embryo.		5
	momento de emergo.		
Q21.	a) 23 Pairs/ 46 chromosomes	1	
	Male: Two types - X, Y	1⁄2	
	Female: One type - X, X	1⁄2	
	b) Acquired Trait Inherited Trait		
	Changes or characters in the non- Changes or characters in the		
	reproductive tissues cannot be reproductive tissues only can be		
	passed on to the DNA of the germ passed on to the DNA of the germ	1,1	
	cells/ next generation.cells/ next generation.Reason: Change in non-reproductive tissues cannot be passed on to the DNA	1,1	
	of the germ cells.	1	5
		1	
<u>I</u>	1	I	

Q22.	a) $h_0 = 2 \text{ cm}, \qquad f = 16 \text{ cm}, \qquad u = -24 \text{ cm}$		
	$\frac{1}{v} - \frac{1}{u} = \frac{1}{f}$		
	v u j	1⁄2	
	1 1 1		
	$\frac{1}{v} - \frac{1}{(-24)} = \frac{1}{16}$		
		1	
	v = 48  cm	1⁄2	
	$m = \frac{h_i}{h_o} = \frac{v}{u}$		
	$m = \frac{h_o}{h_o} u$	1/2	
		, _	
	$h_{\rm i} = \frac{48}{-24} \times 2 = -4 {\rm cm}$	1/2	
	b) $u = -32 \text{ cm}^{-1}$	, <u>r</u>	
	It means that the object is at 2F, hence the image is also at 2F	1	
	$\therefore$ Size of object =Size of image = 2 cm.	1	5
Q23.	(i) The object is always placed to the left of the mirror. This implies that		
	the light from the object falls on the mirror from the left-hand side		
	(ii) All distances parallel to the principal axis are measured from the pole of		
	the mirror.		
	(iii) All the distances measured to the right of the origin (along $+ x$ -axis) are		
	taken as positive, while those measured to the left of the origin		
	(along – x-axis) are taken as negative.		
	(iv) Distances measured perpendicular to and above the principal axis (along		
	+ y-axis) are taken as positive.		
	(v) Distances measured perpendicular to and below the principal axis		
	(along –y-axis) are taken as negative.	14 m 4	
	(any four)	¹ / ₂ x 4 ¹ / ₂	
	Convex mirror	72	
	$f = \frac{r}{2} = 15 \mathrm{cm}$	1 /	
	2	1/2	
	• $u = -30 \mathrm{cm}, \qquad f = -20 \mathrm{cm} \qquad v = ?$		
	$\frac{1}{1+1} = \frac{1}{1-1}$	1⁄2	
	$\overline{v}^{+}\overline{u}^{-}\overline{f}$		
	1 1 1		
	-+	1	
	v = -60  cm	1/2	5
Q24.	a) •The angle through which the incident ray deviates from its path after		
	passing through glass prism/ the angle between the extended incident ray		
	and the emergent rays	1	
	• Different colours of light bend through different angles with respect to		
	the incident ray/ have different refractive index/ have different speeds.	1	
	• By placing a second identical prism in inverted position with respect to		
	first prism	1	

	• Essential co 1) Presence	is a natural spectrum appearing in the s onditions - of tiny water droplets in atmosphere. of Sun behind the observer.	ky after a rain shower.	1 1⁄2, 1⁄2	5
		SECTION – B			
	25) b	26) b	27) c		
	28) a	29) d	30) c		
	31) d	32) Give credit to all	33) a	1 × 9	9
Q34.	Vegetable oil, NaOH solution With the help of litmus: The reaction mixture will turn red litmus blue.			¹ / ₂ , ¹ / ₂ 1	2
Q35.	Binary Fission First stage: Elongation of the nucleus Final stage: Formation of two daughter cells each having a nucleus.			1	2
Q36.		n the lens. 9 image will increase. 9 vill decrease.		1 $\frac{1}{2}$ $\frac{1}{2}$	2